Up-conversion luminescence in Er³⁺/Yb³⁺ co-doped LiYF₄ single crystals

JIAZHONG ZHANG, HAIPING XIA^{*}, LI FU, SHUO YANG, YONGZHANG JIANG, XUEMEI GU, JIANLI ZHANG, DONGJIE WANG, HAOCHUAN JIANG^a

Key laboratory of Photo-electronic Materials, Ningbo University, Ningbo, Zhejiang, 315211, China ^aNingbo Institute of Materials Technology and Engineering, the Chinese Academy of Sciences, Ningbo, Zhejiang, 315201, China

The absorption spectra and up-conversion (UC) luminescence of Er^{3+}/Yb^{3+} ions co-doped LiYF₄ single crystals synthesized by vertical Bridgman method were investigated. Up-conversion emissions at 519, 522, 529 nm, and 541, 550 nm, and 649, 655, 669 nm were observed. The yellowish green light can be achieved from 2.14 mol% Er^{3+} and 4.04 mol% Yb^{3+} co-doped LiYF₄ crystal. The mechanisms of UC emissions are proposed based on the pump power dependent UC spectra. A cooperative energy transfer process from excited Yb^{3+} to Er^{3+} in the form of two-photons has been presented in LiYF₄ at room temperature.

(Received July 12, 2014; accepted May 7, 2015)

Keywords: Up-conversion, Crystal growth, LiYF₄ single crystals, Er³⁺, Yb³⁺

1. Introduction

Up-conversion (UC) luminescence also meaning an anti-stokes luminescence, which the emission of a higher energy photon follows the absorption of two or more low energy photons, has attracted wide attention in some areas, such as biosensors, infrared pumped visible eye-safe lasers, optical telecommunication, photovoltaics and display technology during the past decades[1-4]. A lot of materials doped with rare earth ions, which can be used for up-conversion luminescence, have been reported in glasses [5], nanocrystals [6] and ceramics [7], etc.

Among the trivalent rare earth ions, the up-conversion process of Er³⁺ ions sensitized by Yb³⁺ ions from infrared to visible light has been widely studied because of a larger absorption cross section at near infrared (NIR) of Yb³⁺ and the high-efficiency energy transfer from Yb^{3+} to Er³⁺ ions[8]. Some kinds of studies have been reported on the UC emission of Er^{3+}/Yb^{3+} co-doped materials [4-9]. In order to obtain higher UC efficiency, available host materials play another key role for UC light. Actually, owing to the low phonon energy, fluorides are the mostly investigated materials [10]. Among those materials, the $\mathrm{Er}^{3+}/\mathrm{Yb}^{3+}$ co-doped NaYF₄ powder has a high UC emission efficiency due to its lower phonon energy, relatively high optical damage threshold, and low nonlinear refractive indices [9]. Because of the similar ordered structure and rigid crystal lattice in the RYF₄ (R=Li, Na, K, etc.) family crystals, it is proposed that the LiYF₄ single crystals also possess advantages of excellent UC emission for Er³⁺/Yb³⁺, and have high efficiency of UC emission like that of NaYF₄.

Generally speaking, powders have strong scattering and low transmission while glasses or ceramics have low optimum mechanical, chemical properties, and low luminescent efficiency [11], they are deficient to be applied effectively in optical devices. Some recent works focus on the performance of rare earth doped LiYF₄ single crystals on near and mid-infrared luminescence and LED emission [11-15]. Unfortunately, there are scarcely reports on UC emission of Er^{3+}/Yb^{3+} co-doped RYF₄ (R=Li, Na, K, etc.) single crystals due to the difficult growth of the crystals. Since the RYF₄ (R=Li, Na, K, etc.) single crystals surpass their powders in transmission which is induced from the scattering of the nanoparticles and is of excellent thermal and mechanical properties and high luminescent efficiency, they become possible to extend the limitations of their powders, and have benefit in practical applications.

In a previous work, we reported spectral properties and energy transfer progress in Er^{3+}/Yb^{3+} codoped LiYF₄ crystal[16]. In this work, the enhanced up-conversion at visible wavelength of Er^{3+}/Yb^{3+} co-doped LiYF₄ single crystal was exhibited under excitation of 966 nm light, and the UC mechanisms were investigated through the pump power dependent UC emission.

2. Experimental

The raw materials (LiF, YF₃, ErF₃, and YbF₃) with a purity of 99.99% were chosen to grow the fluoride single crystals. We prepared the single crystals with a molar ratio of LiF: YF₃: ErF₃=51.5:47.5:1.0 (A) and LiF: YF₃: ErF₃: YbF₃= 51.5:45.5:1.0:2.0 (B), in which the rare earths of ErF₃ and YbF₃ were doped into the materials to displace

the location of YF₃. Temperature gradient of the solidliquid interface is about 50-60 °C, and the growth speed is about 0.1-0.15mm/h. The specific details and processes for crystal growth were described in Ref. [15]. The as grown single crystals were cut into small pieces and wellpolished to about 2.2 mm thickness as inserted in Fig. 1 for optical measurements.

The structures of LiYF₄ single crystals was tested by a XD-98X diffract meter (XD-3, Beijing), with CuK α radiation at 0.15403 nm, and the scanning 2 θ was from 10° to 90° with 0.02° increments and 6 s swept time. The absorption spectra from 350 nm to 1100 nm were recorded with a Cary 500 UV/VIS/NIR spectrophotometer (Agilent Co., America). The fluorescence was measured using a FLSP 920 type spectrometer (Edinburgh Co., England). The concentrations of Er³⁺ and Yb³⁺ were measured through ICP (Inductive Coupled Plasma Emission Spectrometer). In (A), the concentration of Er³⁺ is 2.12 mol%; in (B), Er³⁺ and Yb³⁺ are 2.14 mol% and 4.04 mol%, respectively. All the measurements were carried out at room temperature.

3. Results and discussion



Fig. 1. X-ray diffraction patterns of Er^{3+}/Yb^{3+} co-doped LiYF₄ single crystal. Insert is the photo of polished Er^{3+}/Yb^{3+} codoped LiYF₄ single crystal.



Fig. 2. The absorption spectra of Er^{3+} single doped and Er^{3+}/Yb^{3+} co-doped LiYF₄ single crystals.

Fig. 1 shows the XRD pattern of the grown single crystal. The diffraction peak positions of the crystal match perfectly with those of LiYF₄ in the JCPD card (77-0816), one can deduce that the grown crystal is a single orthorhombic phase, which is the scheelite (CaWO₄) structure, and the lattice constants can be calculated from XRD to be a = b = 5.177 Å, c = 10.783 Å.

Fig. 2 shows the absorption spectra of Er^{3+} singly doped and Er^{+3}/Yb^{+3} co-doped LiYF₄ single crystals at room temperature. Through the absorption spectra, we can find that there are some absorption peaks among the visible lights and a wide absorption peak from 940 nm to 1000 nm peaked at 966 nm. The corresponding energy levels were also labeled in Fig. 2. The wide band is attributed to the overlap of the ${}^{2}F_{5/2}$ energy level of Yb³⁺ ions and ${}^{4}I_{11/2}$ of Er^{3+} ions. It is very beneficial to the output of up-conversion luminescence under 966 nm excitation.



Fig. 3. Up-conversion luminescence spectra of Er^{3+} and Er^{3+}/Yb^{3+} co-doped LiYF₄ single crystals under 966 nm excitation.

Fig. 3 shows the comparison of up-conversion luminescence spectra of Er³⁺ singly doped and Er³⁺/Yb³⁺ co-doped LiYF₄ single crystals in the range of 450-700 nm under 966 nm laser irradiation. It can be observed 519, 522, 529, 541, 550, 649, 655, and 669 nm emissions for both singly and double doped samples. The 519, 522, and 529 nm green emissions are attributed to the ${}^{2}H_{11/2} \rightarrow {}^{4}I_{15/2}$ of Er^{3+} ions; 541 and 550 nm green emissions are to ${}^{4}S_{3/2} \rightarrow {}^{4}I_{15/2}$; 649, 655, and 669 nm red emissions are to ${}^{4}F_{9/2} \rightarrow {}^{4}I_{15/2}$. By comparison, the up-conversion emissions intensity of singly and double doped samples, the Er^{3+}/Yb^{3+} co-doped LiYF₄ crystal appears much stronger than that of Er³⁺ singly doped LiYF₄, one can deduce that the introduction of Yb^{3+} can enhance the up conversion emission effectively. Due to the good energy match between ${}^{4}I_{11/2}$ and ${}^{4}F_{7/2}$ of Er^{3+} ions and ${}^{2}F_{5/2}$ of Yb^{3+} ions, when the Er^{3+}/Yb^{3+} co-doped LiYF₄ single crystal is excited by 966 nm light, the energy of Yb³⁺ ions absorbed from 966 nm light can transfer to Er^{3+} ions efficiently. The CIE chromaticity coordinates of the Er^{3+}/Yb^{3+} co-doped

LiYF₄ single crystal is shown in Fig.4. The CIE coordinates (x, y) of the crystal is (0.269, 0.6909) and its color temperature Tc = 6128 K, which is yellowish green color. As we all know that the yellowish green color is sensitive to human eyes, this performance gives possibility for electro-optical devices.



Fig. 4. CIE chromaticity coordinates of 2.14mol% Er³⁺ and 4.04mol% Yb³⁺ co-doped LiYF4 crystal.



Fig. 5. Log-log plot of the up-conversion intensities of Er^{3+}/Yb^{3+} co-doped LiYF4 single crystal on around 529,550, 669 nm emission at the excitation power of 966 nm.

In order to know the UC dynamics of $\text{Er}^{3+}/\text{Yb}^{3+}$ ions co-doped LiYF4 single crystals, the log–log plot of the upconversion intensities of $\text{Er}^{3+}/\text{Yb}^{3+}$ co-doped LiYF₄ single crystal on green and red emission at the excitation power of 966 nm had also been measured and shown in Fig. 5. It has been confirmed that the relation between the emission intensity I_{em} and the infrared power of excitation intensity P is $I_{em} \propto p^n$ in the simplest UC processes, where the n is the number of pump photons required exciting the emitting state, which is indicated by the slope of the luminescence intensity versus pump power in double-logarithmic coordinate [16]. Generally, a realistic UC system that produces detectable UC luminescence will exhibit an intensity-versus-power dependence, which is less than P^n [17]. It can be seen from Fig. 5 the slopes of the luminescence intensity versus pump power in double-logarithmic coordinate for around 529, 550 and 669 nm emissions are 1.986, 1.870, and 1.698, respectively. The nearly square dependences indicate that the up-conversion process to arrive the ${}^{2}H_{11/2}$ and ${}^{4}S_{3/2}$ or ${}^{4}F_{9/2}$ levels need two-photon, and it tally with theory very well.



Fig. 6. Energy level diagram of Er^{3+}/Yb^{3+} *co-doped* $LiYF_4$ *single crystal.*

The up-conversion luminescence possible mechanisms of Er³⁺/Yb³⁺ co-doped materials have been reported elsewhere [19] [20]. Here we try to quantitatively investigate the energy transfer between Yb³⁺ and Er³⁺ under excitation by 966 nm light. Based on the analysis above, UC luminescence mechanisms of Er³⁺/Yb³⁺ codoped LiYF₄ are proposed and shown in Fig. 6. Pumping at 966 nm, the energy of pumping light is mainly absorbed by Yb^{3+} ions via $^2F_{7/2}$ to $^2F_{5/2}.$ Due to the good energy match, it is easily taken place that energy transfers from excited Yb^{3+} ions to their nearby Er^{3+} ions. One excited Yb³⁺ ion can transfer their energy to an Er³⁺ ion and excites the ground-state ${}^{4}I_{15/2}$ to excited ${}^{4}I_{11/2}$ (ET1), or two excited Yb^{3+} ions can transfer their energy to an Er^{3+} ion and excite the ground-state ${}^{4}I_{15/2}$ to excited ${}^{4}F_{7/2}$ (ET2). Meanwhile, the Er^{3+} ions at the excited ${}^{4}I_{11/2}$ state can absorb a photo of 966 nm pump light or energy transfer from Yb³⁺ ions and populate to ⁴F_{7/2} state. Then, the Er³⁺ ions at ${}^{4}F_{7/2}$ state can relax nanradiatively to the ${}^{2}H_{11/2}$ and ${}^{4}S_{3/2}$ state, leading to the green ${}^{5}H_{11/2} \rightarrow {}^{4}I_{15/2}$ and ${}^{4}S_{3/2} \rightarrow {}^{4}I_{15/2}$ emissions. The excited Er^{3+} ions can farther relax nanradiatively to ${}^{4}I_{9/2}$ state; also, the Er³⁺ ions in the ${}^{4}I_{11/2}$ state can relax nanradiatively to the ${}^{4}I_{13/2}$ state and then populate to ${}^{4}F_{9/2}$ state via absorb a phonon of 966 nm pumping light or energy transfer from Yb^{3+} ions, and leading to the red ${}^{4}F_{9/2} \rightarrow {}^{4}I_{15/2}$ emissions.

4 Conclusions

The enhanced up-conversion green and red lights can be obtained in the ${\rm Er}^{3+}/{\rm Yb}^{3+}$ co-doped LiYF₄ single crystal grown by a modified Bridgman method under excitation by a 966 nm diode laser. The UC emissions of the green and red lights arise from two-photon process from excited Yb^{3+} to Er^{3+} energy transfer through the investigation of the pump power dependent UC spectra . The UC green and red lights can be combined to a yellowish green light which is very sensitive to eyes. Such ${\rm Er}^{3+}/{\rm Yb}^{3+}$ co-doped LiYF₄ single crystals may have potential application in the fields of biosensors, infrared pumped visible eye-safe lasers, optical telecommunication and displaying devices.

Acknowledgments

This work was supported in part by the National Natural Science Foundation of China under Grant Nos. 51472125 and 51272109, and by the K. C. Wong Magna Fund in Ningbo University.

References

- J. Zhao, D. Jin, E.P. Schartner, Y. Lu, Y. Liu, A. V. Zvyagin, L. Zhang, J. M. Dawes, P. Xi, J. A. Piper, E. M. Goldys, T. M. Monro, Nature nanotechnology. 8, 729 (2013).
- [2] K. Riwotzki, H. Meyssamy, H. Schnablegger, A. Kornowski, M. Haase, Angewandte Chemie International Edition. 40, 573 (2001).
- [3] C. Jia, L. Sun, Z. Yan, Y. Pang, S. Lü, C. Yan, European Journal of Inorganic Chemistry. 2010, 2626 (2010).
- [4] X. Xie, X. Liu, Nature materials. 11, 842 (2012).
- [5] M. Y. Yoo, J. H. Lee, H. M. Jeong, K. S. Lim, P. Babu, Optical Materials. 35, 1922 (2013).

- [6] N. Niu, P. Yang, Y. Liu, C. Li, D. Wang, S. Gai, F. He, Journal of colloid and interface science. 362, 389 (2011).
- [7] X. Hou, S. Zhou, T. Jia, H. Lin, H. Teng, Physica B: Condensed Matter. 406, 3931 (2011).
- [8] A. S. S. de Camargo, L. A. O Nunes, J. F. Silva, A. C. F. M. Costa, B S Barros, J. E. C. Silva, G. F. de Sá, S. Alves Jr, Journal of Physics: Condensed Matter. 19, 246209 (2007).
- [9] H. Liang, G. Chen, L. Li, Y. Liu, F. Qin, Z. Zhang, Opt. Commun. 282, 3028 (2009).
- [10] H. Zheng, B. Chen, H. Yu, J. Zhang, J. Sun, X. Lia, M. Sun, B. Tian, S. Fu, H. Zhong, B. Dong, R. Hua, H. Xia, Journal of Colloid and Interface Science. 420, 27 (2014).
- [11] L. Fu, H. Xia, Y. Dong, S. Li, H. Jiang, B. Chen, Photonics Journal, IEEE. 6, 1 (2014).
- [12] A. M. E. Santo, A. F. H. Librantz, L. Gomes, P. S. Pizani, I. M. Ranieri, N. D. Vieira Jr., S. L. Baldochi, Journal of Crystal Growth. 292, 149 (2006).
- [13] S. Li, P. Wang, H. Xia, J. Peng, L. Tang, Y. Zhang, H. Jiang, Chinese Optics Letters. 12, 021601 (2014).
- [14] L. Tang, H. Xia, P. Wang, J. Peng, Y. Zhang, H. Jiang, H. Chen, Materials Letters. 104, 37 (2013).
- [15] X. Zhuang, H. Xia, H. Hu, J. Hua, P. Wang, J. Peng, Y. Zhang, H. Jiang, B. Chen, Materials Science and Engineering: B. 178, 326 (2013).
- [16] P. Wang, H. Xia, J. Peng, Y. Hu, T. Tang, Y. Zhang, B. Chen, H. Jiang, Optoelectronics Letters, 9, 4 (2013).
- [17] J. F. Suyver, A. Aebischer, S. García-Revilla, P. Gerner, H. U. Güde, Phys. Rev. B. 71, 125123 (2005).
- [18] M. Pollnau, D. R. Gamelin, S. R. Lüthi, H.U. Güdel, Phys. Rev. B. 61, 3337 (2000).
- [19] X. Mateos, M. C. Pujol, F. Güell, R. Solé, Jna. Gavaldà, J. Massons, M. Aguiló, F. Díaz, Optical Materials. 27, 475 (2004).
- [20] J. P. Wittke, I. Ladany, P. N. Yocom, J. Appl. Phys. 43, 595 (1972).

^{*}Corresponding author: hpxcm@nbu.edu.cn